

Tetrahedron Letters 43 (2002) 4187-4189

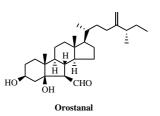
The first stereoselective synthesis of orostanal, a novel abeo-sterol inducing apoptosis in leukemia cells

Bo Liu and Weishan Zhou*

Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai 200032, PR China Received 4 March 2002; revised 11 April 2002; accepted 17 April 2002

Abstract—An efficient approach to the synthesis of a sterol derivative, orostanal, has been achieved by a series of reactions from methyl hyodeoxycholanate. © 2002 Elsevier Science Ltd. All rights reserved.

Recently a novel sterol named as orostanal which consisted of 6-5-6-5 fused rings was isolated from a Japanese marine sponge of *Stelletta hiwasaensis*.¹ Orostanal induced apoptosis in human acute promyelotic leukemia cells at 10 μ g/ml and inhibited 50% cell growth at 1.7 μ M. Some well-known natural anticancer drugs, such as etoposide and camptothecin, also induce apoptosis in cancer cells, so that substances which can induce apoptosis may be useful for human cancer chemotherapy.²



Although it showed meaningful biological activity, only 3.6 mg of orostanal was obtained from 2.0 kg of *S. hiwasaensis.* Thus, it was desirable to establish an efficient route for the stereoselective synthesis of orostanal.

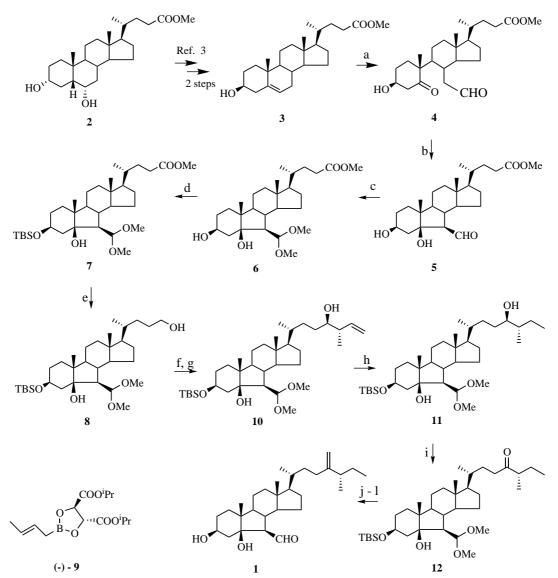
This was achieved starting from 3 which was prepared from hyodeoxycholic acid methyl ester.³ Ozonolysis of 3 in a mixed solvent $(CH_2Cl_2: MeOH =$ 4:1) at -78°C afforded 4 which was converted to 5 in the presence of neutral alumina⁴ at room temperature in a 80% yield, far superior to the yield obtained by Higuchi et al.¹ The aldehyde 5 was protected as its methyl acetal with NH₄Cl as catalyst⁵ to give 6, the secondary hydroxyl at C-3 of which was then silvlated⁶ by TBSCl to yield 7. Next, we planned to reduce the methyl ester moiety on the terminal of the side chain of 7 to -OH by LiAlH₄. However, the TBS ether on A ring of 7 was cleaved to the hydroxyl group when the ester group was reduced. Using LiBH₄⁷ as reductant afforded the desired alcohol 8 which was oxidized⁸ to an aldehyde and coupled with boronate (-)-9 to form 10 under Roush's conditions.⁹ Reduction of **10** by diimide,¹⁰ generated in situ, afforded 11 which was subjected to a Swern oxidation⁷ to yield the ketone **12** (Scheme 1) Considering that a Wittig reaction of 12 might cause racemization¹¹ at C-25, we applied the combined reagent TiCl₄-Zn-CH₂Br₂ for methylenation of the carbonyl group of 12 according to Lombardo's method.¹² After removal of the protective groups under mild conditions,^{13,14} orostanal was obtained as an colorless oil whose spectroscopic data¹⁵ were identical to those reported.¹

To summarize, orostanal was synthesized for the first time via 12 steps in 18% total yield from 3. This synthetic route could provide enough orostanal to probe efficiently its mode of action of apoptosis.

Keywords: stereoselective synthesis; sterol; apoptosis.

^{*} Corresponding author. Tel.: 86-21-64163300; fax: 86-21-64166128; e-mail: zhws@pub.sioc.ac.cn

^{0040-4039/02/\$ -} see front matter @ 2002 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(02)00757-8



Scheme 1. Reagents and conditions: (a) O_3 , CH_2Cl_2 -MeOH (4:1), -78°C, then Me_2S , 90%; (b) neutral Al_2O_3 , benzene, rt, 80%; (c) NH₄Cl, HC(OMe)₃, absolute MeOH, 30°C, 83%; (d) TBSCl, imidazole, DMF, rt, 92%; (e) LiBH₄, THF, reflux, 85%; (f) (COCl)₂, DMSO, CH_2Cl_2 , -78°C, then Et_3N ; (g) (-)-9, toluene, 4 Å MS, -78°C, 92% for two steps; (h) H_2NNH_2 , H_2O_2 , EtOH, 94%; (i) (COCl)₂, DMSO, CH_2Cl_2 , -78°C, then Et_3N , 88%; (j) TiCl₄-Zn-CH₂Br₂ (Lombardo's reagent, prepared according to Ref. 12), CH₂Cl₂, 25°C; (k) LiBF₄, MeCN/CH₂Cl₂ (2:1), trace of water, 25°C; (l) HF, MeCN, 25°C, 50% for three steps.

References

- Miyamoto, T.; Kodama, K.; Aramaki, Y.; Higuchi, R.; Soest, R. W. M. *Tetrahedron Lett.* 2001, 42, 6349–6351.
- 2. Kaufmann, S. H. Cancer Res. 1989, 49, 5870-5878.
- Bharucha, K. R.; Buckley, G. C.; Cross, C. K.; Rubin, L. J.; Ziegler, P. Can. J. Chem. 1956, 34, 982–990.
- 4. Tanabe, K.; Morisawa, Y. Chem. Pharm. Bull. 1963, 11, 536–538.
- Nagata, W.; Wakabayyashi, T.; Narisada, M.; Hayase, Y.; Kamata, S. J. Am. Chem. Soc. 1971, 93, 5740–5785.
- Corey, E. J.; Venkateswarlu, A. J. Am. Chem. Soc. 1972, 94, 6190–6191.
- 7. (a) For preparation of LiBH₄, see: Brown, H. C.; Choi, Y. M.; Narasimhan, S. *Inorg. Chem.* 1982, 21, 3657–3661;
 (b) For reduction of esters with LiBH₄, see: Nystrom, R. F.; Chaikin, S. W.; Brown, W. G. J. Am. Chem. Soc. 1949, 71, 3245–3246.

- 8. Roush, W. R. J. Am. Chem. Soc. 1980, 102, 1390-1404.
- (a) Roush, W. R.; Ando, K.; Powers, D. B.; Palkowitz, A. D.; Halterman, R. L. J. Am. Chem. Soc. 1990, 112, 6339–6348; (b) Roush, W. R.; Halterman, R. L. J. Am. Chem. Soc. 1986, 108, 294–296.
- Paquette, L. A.; Browne, A. R.; Chamot, E.; Blout, J. F. J. Am. Chem. Soc. 1980, 102, 643–651.
- For example, see: (a) Lajunen, M. *Tetrahedron* 1994, *50*, 13181–13198; (b) Frnguelli, F.; Minuti, L.; Taticchi, A. *Synth. Commun.* 1990, *20*, 2507–2517.
- 12. (a) Lombardo, L. *Tetrahedron Lett.* 1982, 23, 4293–4296;
 (b) Lombardo, L. Org. Synth. 1987, 65, 81–89.
- For use of LiBF₄ to deprotect aldehydes from acetals, see: Lipshutz, B. H.; Harvey, D. F. Synth. Commun. 1982, 12, 267–277.
- For use of HF to recover hydroxyl groups from TBS ethers, see: Newton, R. F.; Reynolds, D. P. *Tetrahedron Lett.* 1979, 20, 3981–3982.

Our synthetic orostanal was characterized by ¹H, ¹³C NMR, NOESY and IR spectra. ¹H NMR (600 MHz, CDCl₃): 0.72 (3H, s, Me-18), 0.83 (3H, t, 7.2, Me-29), 0.93 (3H, s, Me-19), 0.94 (3H, d, 6.6, Me-21), 1.00 (3H, d, 6.6, Me-27), 1.73 (1H, dd, 3.1, 14.9, H-4), 2.24 (1H, dd, 2.8, 9.0, H-7), 4.12 (1H, m, H-3), 4.69 (2H, s, CH₂-28), 9.71 (1H, d, 2.8, H-6); ¹³C NMR (150 MHz, CDCl₃): 11.9 (C-29),

12.5 (C-18), 18.4 (C-19), 18.8 (C-21), 19.8 (C-27), 21.6 (C-11), 24.6 (C-15), 26.8 (C-1), 28.0 (C-2), 28.3 (C-16, 26), 30.4 (C-23), 34.6 (C-22), 35.6 (C-20), 39.8 (C-12), 40.0 (C-8), 41.7 (C-25), 44.3 (C-4), 44.8 (C-13), 45.5 (C-10), 50.6 (C-9), 55.6 (C-17), 56.2 (C-14), 64.0 (C-7), 67.4 (C-3), 84.3 (C-5), 107.2 (C-28), 155.2 (C-24), 204.6 (C-6); IR (film, cm⁻¹): 3443, 2928, 2872, 2734, 1720, 1641, 1459, 1378, 887.